

Handbook Of Chemistry And Physics A Ready Reference Book Of Chemical And Physical Data Thirty Sixth Edition 1954 1955

For more than 90 years, researchers around the world have relied on the CRC Handbook of Chemistry and Physics for authoritative, up-to-date data. This year will be no exception. Many of the most heavily used tables in the book receive major updates and expansions, most notably: Physical Properties of Inorganic Compounds - Features nearly 25% more compounds Enthalpy of Fusion - Contains updated values and 20% more compounds, especially inorganics Bond Dissociation Energies - Includes 70% more compounds, including for the first time more than 1200 molecular ions Table of the Isotopes - Brought up to date with research results through the year 2005 Inorganic Ion and Ligand Nomenclature - Incorporates new rules from IUPAC for systematic names Chemical Carcinogens - Updated in accordance with the recent report from the National Toxicology Program Global Temperature Trends - Traces the rise in mean global temperature for the last 150 years New references will also help keep readers up to date.

Get a FREE first edition facsimile with each copy of the 85th! Researchers around the world depend upon having access to authoritative, up-to-date data. And for more than 90 years, they have relied on the CRC Handbook of Chemistry and Physics for that data. This year is no exception. New tables, extensive updates, and added sections mean the Handbook has again set a new standard for reliability, utility, and thoroughness. This edition features a Foreword by world renowned neurologist and author Oliver Sacks, a free facsimile of the 1913 first edition of the Handbook, and thumb tabs that make it easier to locate particular data. New tables in this edition include: Index of Refraction of Inorganic Crystals Upper and Lower Azeotropic Data for Binary Mixtures Critical Solution Temperatures of Polymer Solutions Density of Solvents as a Function of Temperature By popular request, several tables omitted from recent editions are back, including Coefficients of Friction and Miscibility of Organic Solvents. Ten other sections have been substantially revised, with some, such as the Table of the Isotopes and Thermal Conductivity of Liquids, significantly expanded. The Fundamental Physical Constants section has been updated with the latest CODATA/NIST values, and the Mathematical Tables appendix now features several new sections covering topics that include orthogonal polynomials Clebsch-Gordan coefficients, and statistics.

The latest edition of the world's most popular scientific reference features new tables and reference sections on everything from aqueous solubility of organic compounds to flash point data of common substances. Along with the very latest facts and figures, the CRC Handbook of Chemistry and Physics also contains all of the most frequently used data in science, including the periodic table of the elements, basic constants and units, and geophysical data.

CRC Handbook of Chemistry and Physics, 93rd Edition CRC Press

This student edition features over 50 new or completely revised tables, most of which are in the areas of fluid properties and properties of solids. The book also features extensive references to other compilations and databases that contain additional information.

Proudly serving the scientific community for over a century, this 95th edition of the CRC Handbook of Chemistry and Physics is an update of a classic reference, mirroring the growth and direction of science. This venerable work continues to be the most accessed and respected scientific reference in the world. An authoritative resource consisting of tables of data and current international recommendations on nomenclature, symbols, and units, its usefulness spans not only the physical sciences but also related areas of biology, geology, and environmental science. The 95th Edition of the Handbook includes 22 new tables and major updates and expansions. A new series highlighting the achievements of some of the major historical figures in chemistry and physics was initiated with the 94th edition. This series is continued with this edition, which is focused on Galileo Galilei, James Clerk Maxwell, Marie Sklodowska Curie, and Linus Carl Pauling. This series, which provides biographical information, a list of major achievements, and notable quotations attributed to each of the renowned chemists and physicists, will be continued in succeeding editions. Each edition will feature two chemists and two physicists. Available in traditional print format, as an eBook, and online, this reference puts physical property data and mathematical formulas used in labs and classrooms every day within easy reach. New tables: Section 8: Analytical Chemistry Figures of Merit Common Symbols Used in Gas and Liquid Chromatographic Schematic Diagrams Varieties of Hyphenated Gas Chromatography with Mass Spectrometry Section 15: Practical Laboratory Data Standard Fittings for Compressed Gas Cylinders Plug and Outlet Configurations for Common Laboratory Devices Section 16: Health and Safety Information Abbreviations Used in the Assessment and Presentation of Laboratory Hazards Incompatible Chemicals Explosion (Shock) Hazards Water-Reactive Chemicals Testing Requirements for Peroxidizable Compounds Tests for the Presence of Peroxides Pyrophoric Compounds - Compounds That Are Reactive with Air Flammability Hazards of Common Solvents Selection of Laboratory Gloves Selection of Respirator Cartridges and Filters Selection of Protective Laboratory Garments Protective Clothing Levels Chemical Fume Hoods and Biological Safety Cabinets Gas Cylinder Safety and Stamped Markings Laser Hazards in the Laboratory General Characteristics of Ionizing Radiation for the Purpose of Practical Application of Radiation Protection Radiation Safety Units Significantly updated and expanded tables: Section 1: Basic Constants, Units, and Conversion Factors Update of Standard Atomic Weights (2013) Update of Atomic Masses and Abundances Section 8: Analytical Chemistry Expansion of Abbreviations and Symbols Used in Analytical Chemistry Section 9: Molecular Structure and Spectroscopy Update of Bond Dissociation Energies Section 12: Properties of Solids Major update and Expansion of Electron Stopping Powers Section 14: Geophysics, Astronomy, and Acoustics Major Update of Interstellar Molecules Update of Atmospheric Concentration of Carbon Dioxide, 1958-2013 Update of Global Temperature Trend, 1880-2013 Section 15: Practical Laboratory Data Major update of Reference Points on the ITS-90 Temperature Scale Update of Laboratory Solvents and Other Liquid Reagents Section 16: Health and Safety Information Update of Flammability of Chemical Substances Update of Threshold Limits for Airborne Contaminants to 2013 values Appendix B: Update of Sources of Physical and Chemical Data

The CRC Handbook of Chemistry and Physics, 98th Edition is an update of a classic reference. The 98th Edition contains several new features including, but not limited to - a major update to the table of isotopes, the first major compilation of high quality data of protein-ligand binding thermodynamics, and an important new collection of NMR data critical for understanding outcomes of organic syntheses. Plus, twelve lists have been updated such as, the physical properties of organic compounds and the latest experimental values of bond dissociation energies. Building on the new feature first introduced in the 94th edition, four historical figures in science will be honored on the end plates.

Mirroring the growth and direction of science for a century, the CRC Handbook of Chemistry and Physics, now in its 92nd edition, continues to be the most accessed and respected scientific reference in the world, used by students and Nobel Laureates. Available in its traditional print format, the Handbook is also available as an innovative interactive product on DVD and online. Among a wealth of enhancements, this edition analyzes, updates, and validates molecular formulas and weights, boiling and melting points, densities, and refractive indexes in the Physical Constants of Organic Compounds Table through comparisons with critically evaluated data from the NIST Thermodynamics Research Center. New Tables: Analytical Chemistry Abbreviations Used In Analytical Chemistry Basic Instrumental Techniques of Analytical Chemistry Correlation Table for Ultraviolet Active Functionalities Detection of Outliers in Measurements Polymer Properties Second Virial Coefficients of Polymer Solutions Updated Tables: Properties of the Elements and Inorganic Compounds Update of the Melting, Boiling, Triple, and Critical Points of the Elements Fluid Properties Major update and expansion of Viscosity of Gases table Major update and expansion of Thermal Conductivity of Gases table Major update of Properties of Cryogenic Fluids Major update of Recommended Data for Vapor-Pressure Calibration Expansion of table on the Viscosity of Liquid Metals Update of Permittivity (Dielectric Constant) of Gases table Added new refrigerant R-1234yf to Thermophysical Properties of Selected Fluids at Saturation table Molecular Structure and Spectroscopy Major update of Atomic Radii of the Elements Update of Bond Dissociation Energies Update of Characteristic Bond Lengths in Free Molecules Atomic, Molecular, and Optical Physics Update of Electron Affinities Update of Atomic and Molecular Polarizabilities Nuclear and Particle Physics Major update of the Table of the Isotopes Properties of Solids Major update and expansion of the Electron Inelastic Mean Free Paths table Update of table on Semiconducting Properties of Selected Materials Geophysics, Astronomy, and Acoustics Update of the Global Temperature Trend table to include 2010 data Health and Safety Information Major update of Threshold Limits for Airborne Contaminants The Handbook is also available as an eBook.

This is a reproduction of a book published before 1923. This book may have occasional imperfections such as missing or blurred pages, poor pictures, errant marks, etc. that were either part of the original artifact, or were introduced by the scanning process. We believe this work is culturally important, and despite the imperfections, have elected to bring it back into print as part of our continuing commitment to the preservation of printed works worldwide. We appreciate your understanding of the imperfections in the preservation process, and hope you enjoy this valuable book. ++++ The below data was compiled from various identification fields in the bibliographic record of this title. This data is provided as an additional tool in helping to ensure edition identification: ++++ Handbook Of Chemistry And Physics: A Ready-reference Pocket Book Of Chemical And Physical Data Chemical Rubber Company The Chemical Rubber Company, 1913 Chemistry; Physics

Mirroring the growth and direction of science for a century, the Handbook, now in its 93rd edition, continues to be the most accessed and respected scientific reference in the world. An authoritative resource consisting tables of data, its usefulness spans every discipline. This edition includes 17 new tables in the Analytical Chemistry section, a major update of the CODATA Recommended Values of the Fundamental Physical Constants and updates to many other tables. The book puts physical formulas and mathematical tables used in labs every day within easy reach. The 93rd edition is the first edition to be available as an eBook.

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Continues to be the most accurate, reliable and current resource available on data needed by chemists, physicists and engineers. It provides wide coverage of data on properties of inorganic and organic compounds. Some of the most heavily used tables were recently updated and expanded including: Physical Properties of Inorganic Compounds; Enthalpy of Fusion; Bond Dissociation Energies; Table of the Isotopes; Inorganic Ion and Ligand Nomenclature; Chemical Carcinogens; and Global Temperature Trends for the past 150 years.

Celebrating the 100th anniversary of the CRC Handbook of Chemistry and Physics, this 94th edition is an update of a classic reference, mirroring the growth and direction of science for a century. The Handbook continues to be the most accessed and respected scientific reference in the science, technical, and medical communities. An authoritative resource consisting of tables of data, its usefulness spans every discipline. Originally a 116-page pocket-sized book, known as the Rubber Handbook, the CRC Handbook of Chemistry and Physics comprises 2,600 pages of critically evaluated data. An essential resource for scientists around the world, the Handbook is now available in print, eBook, and online formats. New tables: Section 7: Biochemistry Properties of Fatty Acid Methyl and Ethyl Esters Related to Biofuels Section 8: Analytical Chemistry Gas Chromatographic Retention Indices Detectors for Liquid Chromatography Organic Analytical Reagents for the Determination of Inorganic Ions Section 12: Properties of Solids Properties of Selected Materials at Cryogenic Temperatures Significantly updated and expanded tables: Section 3: Physical Constants of Organic Compounds Expansion of Diamagnetic Susceptibility of Selected Organic Compounds Section 5: Thermochemistry, Electrochemistry, and Solution Chemistry Update of Electrochemical Series Section 6: Fluid Properties Expansion of Thermophysical Properties of Selected Fluids at Saturation Major expansion and update of Viscosity of Liquid Metals Section 7: Biochemistry Update of Properties of Fatty Acids and Their Methyl Esters Section 8: Analytical Chemistry Major expansion of Abbreviations and Symbols Used in Analytical Chemistry Section 9: Molecular Structure and Spectroscopy Update of Bond Dissociation Energies Section 11: Nuclear and Particle Physics Update of Summary Tables of Particle Properties Section 14: Geophysics, Astronomy, and Acoustics Update of Atmospheric Concentration of Carbon Dioxide, 1958-2012 Update of Global Temperature Trend, 1880-2012 Major update of Speed of Sound in Various Media Section 15: Practical Laboratory Data Update of Laboratory Solvents and Other Liquid Reagents Major update of

Density of Solvents as a Function of Temperature Major update of Dependence of Boiling Point on Pressure Section 16: Health and Safety Information Major update of Threshold Limits for Airborne Contaminants Appendix A: Major update of Mathematical Tables Appendix B: Update of Sources of Physical and Chemical Data

The latest edition of the world's most popular scientific reference features tables and reference sections on everything from the Periodic Table to the bond lengths in organometallic compounds. Featuring the latest facts and figures, the CRC Handbook of Chemistry and Physics contains all the most frequently used data in science. Topics new in the 82nd edition: Redox data on biochemical compounds Surface tension of common compounds with water as a function of concentration Viscosity of carbon dioxide Optical properties of materials such as binary semiconductors Interstellar molecules Radio frequency allocations Tables updated in the 82nd edition: Physical Constants of Inorganic Compounds-expanded by 15% Critical temperatures and pressures of fluids Solubility of organic compounds in water Atomic weights and natural abundances of isotopes-latest official recommendation from IUPAC Polymer nomenclature Threshold limits for airborne pollutants Chemical carcinogens

In a world with access to unlimited amounts of data, how can users who need to make critical scientific and technical decisions find high quality, reliable data? Today, more than ever, the CRC Handbook of Chemistry and Physics remains a hallmark of quality. For over 100 years, the Handbook has provided property data on chemical compounds and all physical particles that have been reported in the literature, carefully reviewed by subject experts. Every year older collections are updated with the latest values and new areas will be added as science progresses. All data are reviewed and evaluated by subject matter experts Chemical names and property units are standardized, and structures are provided for most substances Over 380 property tables included Contains important information on data-related subjects such as chemical and laboratory safety, and nomenclature

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