

Conceptual Chemistry 4th Edition

This book retains the relaxed, easy-to-read style of the previous edition to provide integrated coverage of organic and biochemistry, applications, and tools that foster problem-solving skills. More than any other, this text offers balance in the topics presented, and in its presentation of the subject of chemistry. The authors focus on three main areas to help readers master the core concepts of general chemistry and enhance their problem-solving skills: problem-solving support and active assessment, conceptual development, and visualization of the molecular behavior of matter. For anyone who wants a relaxed, easy-to-read book that emphasizes major topics in chemistry as well as problem-solving techniques.

Since the third edition of this reference was completed, there have been major changes in the global chemical industry. With less emphasis on new processes for making basic chemicals and more emphasis on pollution prevention and waste disposal, petrochemical processes are giving way to biochemical processes. These changes are reflected in the new processes being developed, many of which have their own names. In addition, niche improvements are still being made in petrochemistry, and some of these processes have new names as well. Gathering and defining a large portion of special named processes that may fall outside standard chemical texts or be scattered among industry manuals, *Encyclopedic Dictionary of Named Processes in Chemical Technology, Fourth Edition* provides a single-source reference on an extensive array of named processes. It provides concise descriptions of those processes in chemical technology that are known by special names that are not self-explanatory. While overviews of the chemical technology

industry are present in other books, most of the names defined within this volume are unique to this compilation. This reference includes named processes in current commercial use around the world, processes that have been or are being piloted on a substantial scale, and even obsolete processes that have been important in the past. The length of the dictionary entries reflects their importance and topicality. The text includes references that document the origins of the processes and review the latest developments. Written by a highly experienced and respected author, this user-friendly text is presented in a practical dictionary format that is useful for a broad audience including industrial chemists and engineers.

Un livre instructif, didactique, pédagogique et complet, il aborde toute la matière de physique du programme L1-L2. S'inspirant de la vie quotidienne, l'auteur aborde la physique à partir d'expériences concrètes et des vidéos. Dans ce livre, Paul G. Hewitt traite des principaux fondements de la physique en évoquant tous les grands concepts qui ont révolutionné cette discipline (de la mécanique à la théorie de la relativité). L'objectif de ce livre est de rendre la physique plus attractive en donnant la possibilité aux étudiants de faire un parallèle entre un phénomène physique et un évènement de la vie de tous les jours. L'auteur alimente donc son propos par de nombreuses démonstrations pertinentes et concrètes pour guider l'étudiant et le mener de la mécanique classique à la physique moderne. Pour illustrer son propos il utilise des vidéos accessibles par QR code. Connu pour ses analogies et ses images issues du réel, son objectif est de faire acquérir aux étudiants, de niveau universitaire, des bases solides en physique. Chaque chapitre se termine par des problèmes variés qui permettent aux étudiants de s'exercer et de s'évaluer.

External representations (pictures, diagrams, graphs,

concrete models) have always been valuable tools for the science teacher. This book brings together the insights of practicing scientists, science education researchers, computer specialists, and cognitive scientists, to produce a coherent overview. It links presentations about cognitive theory, its implications for science curriculum design, and for learning and teaching in classrooms and laboratories. The book presents, in a unified manner, various crystallization design methods. It discusses in detail the geometric framework for representing complex phase behavior involving multiple solutes, enantiomers, hydrates, compounds, polymorphs, and solid solutions through visualization of high-dimensional phase diagrams. It also describes how the impact of transport processes is accounted for using kinetically controlled process paths.

Readers continue to turn to Klein's Organic Chemistry As a Second Language: Second Semester Topics, 4th Edition because it enables them to better understand fundamental principles, solve problems, and focus on what they need to know to succeed. The fourth edition explores the major principles in the field and explains why they are relevant. It is written in a way that clearly shows the patterns in organic chemistry so that readers can gain a deeper conceptual understanding of the material. Topics are presented clearly in an accessible writing style along with numerous hands-on problem solving exercises.

For courses in Thermodynamics. A visual, conceptual and contemporary approach to Physical Chemistry Engel and Reids Thermodynamics, Statistical Thermodynamics, and Kinetics provides a contemporary, conceptual, and visual introduction to physical chemistry. The authors emphasise the vibrancy of physical chemistry today and illustrate its relevance to the world around us using modern applications drawn from biology, environmental science, and material

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science. The 4th Edition provides visual summaries of important concepts and connections in each chapter, offers students just in time math help, and expands content to cover science relevant to physical chemistry.

Conceptual Chemistry Volume I For Class XI

Organic Chemistry, 4th Edition provides a comprehensive yet accessible treatment of all the essential organic chemistry concepts covered in a two-semester course. Presenting a skills-based approach that bridges the gap between organic chemistry theory and real-world practice, Dr. David Klein makes content comprehensible to students while placing special emphasis on developing their problem-solving skills through applied exercises and activities. This edition is available with the new and improved WileyPLUS—an immersive online environment packed with interactive study tools, strategies, and resources that support different learning styles. Organic Chemistry incorporates Klein's acclaimed SkillBuilder program which supplies a wealth of opportunities for students to develop the key skills necessary to succeed in organic chemistry. Each SkillBuilder contains a solved problem that demonstrates a skill and several practice problems of varying difficulty levels—including conceptual and cumulative problems that challenge students to apply the skill in a slightly different environment. An up-to-date collection of literature-based problems exposes students to the dynamic and evolving nature of organic chemistry and its active role in addressing global challenges. Throughout the text, numerous hands-on activities and real-world examples help students understand both the "why" and

the "how" behind organic chemistry.

Conceptual Chemistry Understanding Our World of Atoms and Molecules Prentice Hall

Presents the entire inorganic field as a logical development of basic ideas, incorporating significant early contributions, factual data, and the resulting modern ideas involving the scope and significance of inorganic chemistry. Proceeds sensibly from the origins of the elements through atomic structure, molecular structure, bonding and properties related to bonding, and reactions considered by types, conditions, and mechanisms.

This new edition of Robert G. Mortimer's Physical Chemistry has been thoroughly revised for use in a full year course in modern physical chemistry. In this edition, Mortimer has included recent developments in the theories of chemical reaction kinetics and molecular quantum mechanics, as well as in the experimental study of extremely rapid chemical reactions. While Mortimer has made substantial improvements in the selection and updating of topics, he has retained the clarity of presentation, the integration of description and theory, and the level of rigor that made the first edition so successful. * Emphasizes clarity; every aspect of the first edition has been examined and revised as needed to make the principles and applications of physical chemistry as clear as possible. * Proceeds from fundamental principles or postulates and shows how the consequences of these principles and postulates apply to the chemical and physical phenomena being studied. * Encourages the student not only to know the applications

in physical chemistry but to understand where they come from. * Treats all topics relevant to undergraduate physical chemistry.

Part I: Process design -- Introduction to design -- Process flowsheet development -- Utilities and energy efficient design -- Process simulation -- Instrumentation and process control -- Materials of construction -- Capital cost estimating -- Estimating revenues and production costs -- Economic evaluation of projects -- Safety and loss prevention -- General site considerations -- Optimization in design -- Part II: Plant design -- Equipment selection, specification and design -- Design of pressure vessels -- Design of reactors and mixers -- Separation of fluids -- Separation columns (distillation, absorption and extraction) -- Specification and design of solids-handling equipment -- Heat transfer equipment -- Transport and storage of fluids.

This book is about how students are taught the periodic table. It reviews aspects of the periodic table's development, using the history and philosophy of science. The teaching method presented in this book is ideal for teaching the subject in high school and at introductory university level. Chemistry students taught in this new, experimental way are compared with those taught in the traditional way and the author describes how tests found more conceptual responses from the experimental group than the control group. The historical aspects of importance to this teaching method are: the role of the Karlsruhe Congress of 1860; the accommodation of the chemical elements in the periodic table; prediction of elements that were discovered later;

corrections of atomic weights; periodicity in the periodic table as a function of the atomic theory; and the accommodation of argon. The experimental group of students participated in various activities, including: discussion of various aspects related to the history and philosophy of science; construction of concept maps and their evaluation by the students; PowerPoint presentations; and interviews with volunteer students. The main objective of this monograph is to incorporate history and philosophy of science in the chemistry curriculum in order to provide students an overview of the dynamics of scientific research, which involves controversies, conflicts and rivalries among scientists, that is the humanising aspects of science. A major thesis of this book is the parallel between the construction of knowledge by the students and the scientists. In looking for this relationship, it is not necessary that ontogeny recapitulate phylogeny, but rather to establish that students can face similar difficulties in conceptualising problems as those faced by the scientists in the past. Given the vast amount of literature on students' alternative conceptions (misconceptions) in science, it is plausible to suggest that these can be considered not as mistakes, but rather as tentative models, leading to greater conceptual understanding. Just as scientists resist changes in the 'hard-core' of their beliefs by offering 'auxiliary hypotheses', students may adopt similar strategies. Conceptual change, in science education can thus be conceptualised as building of tentative models that provide greater explanatory power to students' understanding.

Thermodynamics is the science that describes the behavior of matter at the macroscopic scale, and how this arises from individual molecules. As such, it is a subject of profound practical and fundamental importance to many science and engineering fields. Despite extremely varied applications ranging from nanomotors to cosmology, the core concepts of thermodynamics such as equilibrium and entropy are the same across all disciplines. A Conceptual Guide to Thermodynamics serves as a concise, conceptual and practical supplement to the major thermodynamics textbooks used in various fields. Presenting clear explanations of the core concepts, the book aims to improve fundamental understanding of the material, as well as homework and exam performance. Distinctive features include: Terminology and Notation Key: A universal translator that addresses the myriad of conventions, terminologies, and notations found across the major thermodynamics texts. Content Maps: Specific references to each major thermodynamic text by section and page number for each new concept that is introduced. Helpful Hints and Don't Try Its: Numerous useful tips for solving problems, as well as warnings of common student pitfalls. Unique Explanations: Conceptually clear, mathematically fairly simple, yet also sufficiently precise and rigorous. A more extensive set of reference materials, including older and newer editions of the major textbooks, as well as a number of less commonly used titles, is available online at <http://www.conceptualthermo.com>. Undergraduate and graduate students of chemistry, physics, engineering,

geosciences and biological sciences will benefit from this book, as will students preparing for graduate school entrance exams and MCATs.

This loose-leaf, three-hole punched version of the textbook gives students the flexibility to take only what they need to class and add their own notes--all at an affordable price. For courses in Thermodynamics. A visual, conceptual and contemporary approach to Physical Chemistry Engel and Reid's Thermodynamics, Statistical Thermodynamics, and Kinetics provides a contemporary, conceptual, and visual introduction to physical chemistry. The authors emphasize the vibrancy of physical chemistry today and illustrate its relevance to the world around us, using modern applications drawn from biology, environmental science, and material science. The 4th Edition provides visual summaries of important concepts and connections in each chapter, offers students "just-in-time" math help, and expands content to cover science relevant to physical chemistry. Tutorials in Mastering(tm) Chemistry reinforce students' understanding of complex theory in Quantum Chemistry and Thermodynamics as they build problem-solving skills throughout the course. Also available with Mastering Chemistry Mastering(tm) is the teaching and learning platform that empowers you to reach every student. By combining trusted author content with digital tools developed to engage students and emulate the office-hour experience, Mastering personalizes learning and often improves results for each student. Instructors ensure students arrive ready to learn by assigning educationally effective content before class, and

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encourage critical thinking and retention with in-class resources such as Learning Catalytics. Learn more about Mastering Chemistry. NOTE: You are purchasing a standalone product; Mastering(tm) Geography does not come packaged with this content. Students, if interested in purchasing this title with Mastering Geography, ask your instructor to confirm the correct package ISBN and Course ID. Instructors, contact your Pearson representative for more information. If you would like to purchase both the loose-leaf version of the text and Mastering Geography, search for: 0134813790 / 9780134813790 Physical Chemistry: Thermodynamics, Statistical Thermodynamics, and Kinetics, Books a la Carte Plus MasteringChemistry with Pearson eText -- Access Card Package, 4/e

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For courses in Chemistry. Building 21st Century Data Analysis and Problem-Solving Skills in Modern Chemistry The Fourth Edition of Nivaldo Tro's Chemistry: A Molecular Approach reinforces development of 21st century skills including data interpretation and analysis, problem solving and quantitative reasoning, applying conceptual understanding to new situations and peer-to-peer collaboration. Nivaldo Tro presents chemistry visually through multi-level images-macroscopic, molecular, and symbolic representations-helping readers see the connections between the world they see around them (macroscopic), the atoms and molecules that compose the world (molecular), and the formulas they write down on paper (symbolic). The benefits of Dr. Tro's problem-solving approach are reinforced through digital,

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Interactive Worked Examples that provide an office-hour type of environment and expanded coverage on the latest developments in chemistry. New Key Concept Videos explain difficult concepts while new end-of-chapter problems including Group Work questions and Data Interpretation and Analysis questions engage readers in applying their understanding of chemistry. The revision has been constructed to easily incorporate material to engage readers. Also available with MasteringChemistry MasteringChemistry from Pearson is the leading online homework, tutorial, and assessment system, designed to improve results by engaging you before, during, and after class with powerful content. Instructors ensure you arrive ready to learn by assigning educationally effective content before class, and encourage critical thinking and retention with in-class resources such as Learning Catalytics(tm). You can further master concepts after class through traditional and adaptive homework assignments that provide hints and answer-specific feedback. The Mastering gradebook records scores for all automatically graded assignments in one place, while diagnostic tools give instructors access to rich data to assess your understanding and misconceptions. Mastering brings learning full circle by continuously adapting to your learning and making learning more personal than ever-before, during, and after class. Note: You are purchasing a standalone product; MasteringChemistry does not come packaged with this content. Students, if interested in purchasing this title with MasteringChemistry, ask your instructor for the correct package ISBN and Course ID. Instructors,

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A long-established leader in liberal arts chemistry, John Suchocki's *Conceptual Chemistry* continues to take a student-centered approach in the new Fifth Edition. This robust program uses a conversational writing style to engage and involve non-science majors, helping them visualize the behavior of atoms and molecules to understand our macroscopic environment. With a focus on fundamental concepts - not mathematical models. For courses in Thermodynamics. A visual, conceptual and contemporary approach to Physical Chemistry Engel and Reid's *Thermodynamics, Statistical Thermodynamics, and Kinetics* provides a contemporary, conceptual, and visual introduction to physical chemistry. The authors emphasize the vibrancy of physical chemistry today and illustrate its relevance to the world around us, using modern applications drawn from biology, environmental science, and material science. The 4th Edition provides visual summaries of important concepts and connections in each chapter, offers students "just-in-time" math help, and expands content to cover science relevant to physical chemistry. Tutorials in

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Chemistry The Fourth Edition of Nivaldo Tro's Chemistry: A Molecular Approach reinforces students' development of 21st century skills including data interpretation and analysis, problem solving and quantitative reasoning, applying conceptual understanding to new situations and peer-to-peer collaboration. Nivaldo Tro presents chemistry visually through multi-level images—macroscopic, molecular, and symbolic representations—helping students see the connections between the world they see around them (macroscopic), the atoms and molecules that compose the world (molecular), and the formulas they write down on paper (symbolic). The benefits of Dr. Tro's problem-solving approach are reinforced through digital, Interactive Worked Examples that provide students with an office-hour type of environment and expanded coverage on the latest developments in chemistry. New Key Concept Videos explain difficult concepts while new end-of-chapter problems including Group Work questions and Data Interpretation and Analysis questions engage students in applying their understanding of chemistry. The revision has been constructed to easily incorporate material for instructors and students to engage in Before, During, and After class activities. MasteringChemistry™ is not included. Students, if MasteringChemistry is a recommended/mandatory component of the course, please ask your instructor for the correct ISBN and course ID. MasteringChemistry should only be purchased when required by an instructor. Instructors, contact your Pearson representative for more information. MasteringChemistry from Pearson is the

leading online homework, tutorial, and assessment system, designed to improve results by engaging students before, during, and after class with powerful content. Instructors ensure students arrive ready to learn by assigning educationally effective content before class, and encourage critical thinking and retention with in-class resources such as Learning Catalytics™.

The 4th Edition of the book Objective NCERT Xtract -Chemistry for NEET/ JEE Main, Class 11 & 12, AIIMS, BITSAT consists of Quality Selected MCQs as per current NCERT syllabus covering the entire syllabus of 11th and 12th standard. The most highlighting feature of the book is the inclusion of a lot of new questions created exactly on the pattern of NCERT. • This book-cum-Question Bank spans through 30 chapters. • The book provides a detailed 2 page Concept Map for Quick Revision of the chapter. • This is followed by 3 types of objective exercises: 1. Topic-wise Concept Based MCQs 2. NCERT Exemplar & Past JEE Main, BITSAT, NEET & AIIMS Questions 3. 15-20 Challenging Questions in Try If You Can Exercise • Detailed explanations have been provided for all typical MCQs that need conceptual clarity. • The book also includes 5 Mock Tests for Self Assessment. This book assures complete syllabus coverage by means of questions for more or less all significant concepts of Chemistry. In nutshell this book will act as the BEST PRACTICE & REVISION MATERIAL for all PMT/ PET entrance exams.

Physical Chemistry and Its Biological Applications presents the basic principles of physical chemistry and shows how the methods of physical chemistry are being

applied to increase understanding of living systems. Chapters 1 and 2 of the book discuss states of matter and solutions of nonelectrolytes. Chapters 3 to 5 examine laws in thermodynamics and solutions of electrolytes. Chapters 6 to 8 look at acid-base equilibria and the link between electromagnetic radiation and the structure of atoms. Chapters 9 to 11 cover different types of bonding, the rates of chemical reactions, and the process of adsorption. Chapters 12 to 14 present molecular aggregates, magnetic resonance spectroscopy and photochemistry, and radiation. This book is useful to biological scientists for self-study and reference. With modest additions of mathematical material by the teacher, the book should also be suitable for a full-year major's course in physical chemistry. The book is revised specifically to address the needs of the latest course curriculum in Engineering Chemistry for the first semester students of all branches of engineering. The topics covered in the book are customarily taught in several universities and institutes. The book exposes students to fundamental knowledge in Water technology • Applications of surface chemistry and concept of nuclear energy and energy storage devices • Alloys and phase rule • Electrochemistry and principle involved in corrosion and its inhibition and protective coatings • Analysis of fuels and combustion KEY FEATURES • Several worked-out examples to help students reinforce their comprehension of theory • Numerous short and descriptive questions at the end of each chapter to test and foster students' conceptual understanding of the subject • Chapter-end problems to help students become proficient in problem solving TARGET AUDIENCE Students of first-year BE/BTech (All Branches)

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NOTE: This edition features the same content as the traditional text in a convenient, three-hole-punched, loose-leaf version. Books a la Carte also offer a great value--this format costs significantly less than a new textbook. Before purchasing, check with your instructor or review your course syllabus to ensure that you select the correct ISBN. Several versions of Pearson's MyLab & Mastering products exist for each title, including customized versions for individual schools, and registrations are not transferable. In addition, you may need a CourseID, provided by your instructor, to register for and use Pearson's MyLab & Mastering products.

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In recent years, our world has experienced a profound shift and progression in available computing and knowledge sharing innovations. These emerging advancements have developed at a rapid pace, disseminating into and affecting numerous aspects of contemporary society. This has created a pivotal need for an innovative compendium encompassing the latest trends, concepts, and issues surrounding this relevant discipline area. During the past 15 years, the Encyclopedia of Information Science and Technology has become recognized as one of the landmark sources of the latest knowledge and discoveries in this discipline. The Encyclopedia of Information Science and Technology, Fourth

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Edition is a 10-volume set which includes 705 original and previously unpublished research articles covering a full range of perspectives, applications, and techniques contributed by thousands of experts and researchers from around the globe. This authoritative encyclopedia is an all-encompassing, well-established reference source that is ideally designed to disseminate the most forward-thinking and diverse research findings. With critical perspectives on the impact of information science management and new technologies in modern settings, including but not limited to computer science, education, healthcare, government, engineering, business, and natural and physical sciences, it is a pivotal and relevant source of knowledge that will benefit every professional within the field of information science and technology and is an invaluable addition to every academic and corporate library.

Real success in your chemistry course depends on far more than memorizing equations. Introductory Chemistry, Fourth Edition helps you develop a deeper understanding of chemical concepts as well as your problem-solving skills, with a reader-friendly style and stunning illustrations that have made this text a student favorite. The authors' conceptual approach focuses on the concepts behind chemical equations, to help you become a more proficient problem solver. Unlike other books that emphasize rote memorization of problem-solving algorithms, this text helps you master the quantitative skills and understanding you'll need to gain a real understanding of chemistry.

Readers continue to turn to Klein's Organic Chemistry as a Second Language: First Semester Topics, 4th Edition because it enables them to better understand fundamental principles, solve problems, and focus on what they need to know to succeed. This edition explores the major principles in the field and explains why they are relevant. It is written in a

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way that clearly shows the patterns in organic chemistry so that readers can gain a deeper conceptual understanding of the material. Topics are presented clearly in an accessible writing style along with numerous hands-on problem solving exercises.

Science teaching has evolved as a blend of conventional methods and modern aids owing to the changing needs and techniques of education with an objective to develop scientific attitude among the students. This Fourth Edition of Innovative Science Teaching aims to strike balance between modern teaching methods and time-tested theories. FEATURES OF THE FOURTH EDITION • Chapters 3, 8 and 13 have been thoroughly revised and updated in the light of advancements of application of technology in teaching. • Chapter 13—New Technology to Promote Learning—has been expanded to include the impact of technology on teaching and learning. • E-learning materials and website addresses relevant to science teaching have been updated. • All chapters have been revised and extensive coverage of all aspects of modern teaching has been included. This edition of Innovative Science Teaching is designed for the undergraduate and postgraduate students of Education specializing in science teaching. It can also prove useful as a reference book for administrators, researchers and teacher-trainers. TARGET AUDIENCE • B.Ed (specialization in Science Teaching) • M.Ed (specialization in Science Teaching) • Diploma Courses in Education

This book argues that the traditional image of Feyerabend is erroneous and that, contrary to common belief, he was a great admirer of science. It shows how Feyerabend presented a vision of science that represented how science really works. Besides giving a theoretical framework based on Feyerabend's philosophy of science, the book offers criteria that can help readers to evaluate and understand research reported in

important international science education journals, with respect to Feyerabend's epistemological anarchism. The book includes an evaluation of general chemistry and physics textbooks. Most science curricula and textbooks provide the following advice to students: Do not allow theories in contradiction with observations, and all scientific theories must be formulated inductively based on experimental facts. Feyerabend questioned this widely prevalent premise of science education in most parts of the world, and in contrast gave the following advice: Scientists can accept a hypothesis despite experimental evidence to the contrary and scientific theories are not always consistent with all the experimental data. No wonder Feyerabend became a controversial philosopher and was considered to be against rationalism and anti-science. Recent research in philosophy of science, however, has shown that most of Feyerabend's philosophical ideas are in agreement with recent trends in the 21st century. Of the 120 articles from science education journals, evaluated in this book only 9% recognized that Feyerabend was presenting a plurality of perspectives based on how science really works. Furthermore, it has been shown that Feyerabend could even be considered as a perspectival realist. Among other aspects, Feyerabend emphasized that in order to look for breakthroughs in science one does not have to be complacent about the truth of the theories but rather has to look for opportunities to "break rules" or "violate categories." Mansoor Niaz carefully analyses references to Feyerabend in the literature and displays the importance of Feyerabend's philosophy in analyzing, historical episodes. Niaz shows through this remarkable book a deep understanding to the essence of science. - Calvin Kalman, Concordia University, Canada In this book Mansoor Niaz explores the antecedents, context and features of Feyerabend's work and offers a more-nuanced

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understanding, then reviews and considers its reception in the science education and philosophy of science literature. This is a valuable contribution to scholarship about Feyerabend, with the potential to inform further research as well as science education practice.- David Geelan, Griffith University, Australia

'Conceptual Physical Science Explorations' presents a clear and engaging introduction to physics, chemistry, astronomy, and earth sciences. The authors use analogies and examples to clarify key concepts and help students better understand the world around them.

Enhanced with new problems and applications, the Fourth Edition of CHEMISTRY FOR ENGINEERING STUDENTS provides a concise, thorough, and relevant introduction to chemistry that prepares you for further study in any engineering field. Updated with new conceptual understanding questions and applications specifically geared toward engineering, the book emphasizes the connection between molecular properties and observable physical properties and the connections between chemistry and other subjects such as mathematics and physics.

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It goes without saying that atomic structure, including its dual wave-particle nature, cannot be demonstrated in the classroom. Thus, for most science teachers, especially those in physics and chemistry, the textbook is their key resource and their students' core source of information. Science education historiography recognizes the role played by the history and philosophy of science in developing the content of our textbooks, and with this in mind, the authors analyze more than 120 general chemistry textbooks published in the

USA, based on criteria derived from a historical reconstruction of wave-particle duality. They come to some revealing conclusions, including the fact that very few textbooks discussed issues such as the suggestion, by both Einstein and de Broglie, and before conclusive experimental evidence was available, that wave-particle duality existed. Other large-scale omissions included de Broglie's prescription for observing this duality, and the importance of the Davisson-Germer experiments, as well as the struggle to interpret the experimental data they were collecting. Also untouched was the background to the role played by Schrödinger in developing de Broglie's ideas. The authors argue that rectifying these deficiencies will arouse students' curiosity by giving them the opportunity to engage creatively with the content of science curricula. They also assert that it isn't just the experimental data in science that matters, but the theoretical insights and unwonted inspirations, too. In addition, the controversies and discrepancies in the theoretical and experimental record are key drivers in understanding the development of science as we know it today.

New Scientist magazine was launched in 1956 "for all those men and women who are interested in scientific discovery, and in its industrial, commercial and social consequences". The brand's mission is no different today - for its consumers, New Scientist reports, explores and interprets the results of human endeavour set in the context of society and culture. The history of chemistry is a story of human endeavor-and as ever T ratic as human nature itself. Progress has been made in fits and starts, and it has come from all parts of the globe. Because the scope of this history is considerable (some 100,000 years), it is necessary to impose some order, and we have organized the text around three dis cernible-albeit gross--divisions of time: Part 1 (Chaps. 1-7) covers 100,000

