

## Chemistry The Central Science Lab Manual 12th Edition Free

For two-semester general chemistry lab courses Introducing basic lab techniques and illustrating core chemical principles Prepared by John H. Nelson and Kenneth C. Kemp, both of the University of Nevada, this manual contains 43 finely tuned experiments chosen to introduce basic lab techniques and to illustrate core chemical principles. In the 14th Edition, all experiments were carefully edited for accuracy, safety, and cost. Pre-labs and questions were revised and new experiments added concerning solutions, polymers, and hydrates. Each of the experiments is self-contained, with sufficient background material, to conduct and understand the experiment. Each has a pedagogical objective to exemplify one or more specific principles. Because the experiments are self-contained, they may be undertaken in any order, although the authors have found in their General Chemistry course that the sequence of Experiments 1 through 7 provides the firmest background and introduction. The authors have included pre-lab questions to answer before starting the lab. The questions are designed to help in understanding the experiment, learning how to do the necessary calculations to treat their data, and as an incentive for reading the experiment in advance. These labs can also be customized through Pearson Collections, our custom database program. For more information, visit <https://www.pearsonhighered.com/collections/>

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A standard in the industry, this best-selling lab manual was written in conjunction with Brown/LeMay/Bursten's Chemistry: The Central Science, 7/e but can be used as a stand-alone lab manual. This edition has been updated to reflect environmental concerns.

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The acknowledged leader and standard in general chemistry, this book maintains its effective and proven features—clarity of writing, scientific integrity, currency, strong exercises, visual emphasis and consistency in presentation. It offers readers an integrated educational solution to the challenges of the learning with an expanded media program that works in concert with the book, helping them to approach problem solving, visualization, and applications with greater success. Chapter topics cover: Matter and Measurement; Atoms, Molecules, and Ions; Stoichiometry: Calculations with Chemical Formulas and Equations; Aqueous Reactions and Solution Stoichiometry; Thermochemistry; Electronic Structure of Atoms; Periodic Properties of the Elements; Basic Concepts of Chemical Bonding; Molecular Geometry and Bonding Theories; Gases; Intermolecular Forces, Liquids, and Solids; Modern Materials; Properties of Solutions; Chemical Kinetics; Chemical Equilibrium; Acid-Base Equilibria; Additional Aspects of Equilibria; Chemistry of the Environment; Chemical Thermodynamics; Electrochemistry; Nuclear Chemistry; Chemistry of the Nonmetals; Metals and Metallurgy; Chemistry of Coordination Compounds; and The Chemistry of Life: Organic and Biological Chemistry. For individuals interested in the study of general chemistry.

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